Stereocontrolled Synthesis of Enantiomerically Pure Dienyl Sulfoxides via Palladium-Catalyzed Coupling Reactions

Robert S. Paley*, Alfonso de Dios

Swarthmore College, Department of Chemistry, Swarthmore, Pennsylvania 19081

Roberto Fernández de la Pradilla*1

Instituto de Ouímica Orgánica, C.S.I.C., Juan de la Cierva 3, 28006 Madrid, Spain

<u>Summary:</u> The palladium-catalyzed coupling of enantiopure 2-halovinylsulfoxides and (E)-vinyl stannanes proceeds in an efficient, stereospecific manner to afford enantiopure 1-sulfinyldienes.

The use of chiral sulfoxides as a element of enantio- and diastereocontrol has ample precedent in a variety of synthetic processes.²⁻⁵ In the context of a project examining the diastereoselectivity of transition metal catalyzed cycloisomerizations of enantiopure sulfoxides bearing multiple unsaturation, we required a general and flexible stereocontrolled method for the preparation of acyclic 1-sulfinyl dienes under mild conditions. Despite several recent reports⁶ of selective dienyl sulfoxide syntheses, a stereochemically versatile method which would facilitate the preparation of any desired dienyl sulfoxide isomer was not available. In this paper, we report an approach to the synthesis of two of the four possible stereoisomers that is based on the coupling of halovinylsulfoxides with vinyl stannanes, using the methodology pioneered by Stille.⁷

Judging from the availability of enantiopure trans and cis-2-halovinylsulfoxides⁸, we reasoned that the desired 1-sulfinyldienes could be constructed, in a stereocontrolled manner, by taking advantage of wellestablished palladium-catalyzed coupling reactions with vinylic metal species. Because of the expected exceptilicity of boron, aluminum, or zirconium reagents, and their potential incompatibility with the sulfoxide group, we decided to focus our efforts on coupling vinylstannanes to halovinylsulfoxides 1a and 1b. After screening several ligand/catalyst combinations (as well as different reaction solvents and temperatures) for the reaction of trans-2-bromovinylsulfoxide 1a with one equivalent of vinyltributylstannane, it was found that treatment with Pd2(dba)3°CHCl3 (2 mol %) and PPh3 (8 mol %), in refluxing THF, (Method A), cleanly produced the desired dienvisulfoxide 4a in excellent yield. Examination of the ¹H NMR spectrum of the crude reaction mixture at 300 MHz revealed that the product (Table L entry 1) was stereochemically homogeneous: furthermore, its enantiomeric purity was judged to be 100% after analysis of the 1H NMR spectrum of the purified material in the presence of Eu(hfc)3. The generality of the new methodology was tested by performing the coupling reaction with a variety of vinyl stannanes 10 (Table I, entries 2 and 3). It is significant to note that the optical rotations of products 4b and 4c are virtually identical to those reported by Solladié and co-workers control of the control of th for these compounds, synthesized by a different route; this additional evidence is conformation that the optical purity of the sulfoxide is not compromised by the coupling process reported here.

After successfully demonstrating the feasibility of this approach with a trans-bromovinylsulfoxide, we turned our attention towards effecting an analogous transformation using the cis isomers and thus establish a stereocontrolled route to the challenging 1Z-1-sulfinyl dienes. Unfortunately, an initial attempt to couple vinyl-

Table I. Palladium-Catalyzed Synthesis of Enantiomerically Pure Dienvl Sulfoxides¹¹

X
$$P-tol^{\frac{1}{4}}.$$

1a X = Br
1b X = I
$$R^{2}$$

$$R^{1}$$

$$R^{2}$$

$$R^{2}$$

$$R^{3}$$

$$Pd cat.$$

$$R^{1}$$

$$R^{2}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

$$R^{3}$$

$$R^{4}$$

$$R^{2}$$

$$R^{3}$$

$$R^{4}$$

$$R^{5}$$

$$R$$

		vinvlstannanes				1-sulfinyldienes			
 entry	sulfoxide	R	R ¹	R ²	ref.	methoda	product	% viel	d [a]n (c)c
1	1 a	Н	Н	H		A	4a	87	+ 283.8 (0.68)
2	1a	Н	Ph	H	10a	A	4b	87	+ 168.9 (0.87)
	:								+ 225.0 (0.44) ^d
3	1a	H	Me	Me	10c	A	4c	80	+ 218.6 (1.06)d
4	2	H	H	H		В	5a	91	- 438.5 (0.52)
5	2	Me	н	H	10c	В	5b	80-	- 329.5 (0.78)
6	2	H	Me	Mc	10c	В	5c	91	- 503.7 (1.08)
7	3	H	Ph	H	10a	В	5d	92	- 663.2 (0.98)
8	1 b	Me	Н	H	10c	В	4d	83	+ 207.6 (0.66)
9	1 b	EtO	н .	H		В	4e	90	+ 382.8 (0.87)
10	2	EtO	H	Н		В	5e	83	- 623.8 (0.98)
11	1a	H	CH(OEt)2	H	10ь	A	4f	85	+ 142.9 (1.05)
12	· 2	H	CH(OEt)2	Н	10b	В	5f	76	- 244.0 (0.25)
13	1a	H	¢	H	10c	A	4g	83	+ 78.3 (0.97)

^a Reaction conditions: Method A: Vinyl stannane (1.0 eq.), Pd₂dbe₃·CHCl₃ (2 mol %), PPh₃ (8 mol %), refluxing THF, 30 min. Method B: Vinyl stannane (1.0 eq), Pd(CH₃CN)₂Cl₂ (2 mol %), DMF, RT, 3 min to 1 h. ^b Unoptimized yields of pure products isolated by chromatography. ^c Optical rotations measured in CHCl₃ unless otherwise noted. ^d Optical rotation measured in acetone; see text and ref. 6e. ^c Vinyl stannanc 6; See text.

tributylstannane to iodovinylsulfoxide 2 using Method A was nonstereoselective. However, we were pleased to discover that milder reaction conditions, Pd(CH₃CN)₂Cl₂ (2 mol %), in DMF at room temperature ^{7b}, cleanly and rapidly effected the desired transformation. Again, the coupling proceeded stereospecifically, with complete retention of double bond stereochemistry (Table I, entry 4). In addition to coupling various vinyl stannanes to 2 (entries 5 and 6), this procedure (Method B), could also be used with trisubstituted iodovinylsulfoxide 3 to produce an excellent yield of diene 5d (entry 7), stereochemically homogeneous as determined by NOESY experiments. Not unexpectedly, Method B also proved to be practical for couplings with trans-2-

iodovinylsulfoxide <u>1h</u> (entries 8 and 9). Also, the straightforward preparation of oxygenated dienes <u>4e</u> and <u>5e</u> (entries 9 and 10) illustrates the versatility of our methodology.

After establishing the generality of these procedures, we sought to prepare dienyl sulfoxides which could be elaborated in subsequent operations. To this end, dienyl sulfoxides 4f, 4g, and 5f (entries 11-13), each possessing a latent aldehyde, were efficiently prepared from the corresponding vinyl stannanes. Of particular interest is that the precursor of 4g, vinyl stannane 6g, was prepared from the corresponding vinyl bromide by standard halogen-metal exchange methodology 10c, in an 80 % yield, despite the possibility of lithiation of the dithiane ring. The coupling of this new vinyl stannane to bromovinylsulfoxide 1g proceeded uneventfully to produce 1g (eq. 1); the chemistry of the deprotected dienylsulfoxide (i.e., the corresponding g-oxo-1-sulfinyldiene) is currently being explored in our laboratories.

To summarize, enantiopure dienyl sulfoxides are now readily available by virtue of an efficient Stille-type coupling process. In a significant improvement over previous methods, the (E, E) and (Z, E)-1-sulfinyldienes presented here can now be prepared in excellent yield, and in a stereochemically reliable manner, through the use of stereodefined vinyl stannanes. The possibility of extending this methodology to include the preparation of the (E, Z) and (Z, Z)-1-sulfinyldienes is presently being examined, as is the synthesis of the related 2-sulfinyldienes. These results, and those pertaining to the chemistry of the sulfinyldienes in the present study, will be reported in due course.

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REFERENCES:

- On leave, 1992-93. Current address: Massachusetts Institute of Technology, Department of Chemistry, Cambridge, MA 02139-4307.
- For reviews, see: (a) The Chemistry of Sulphones and Sulphoxides; Patai, S.; Rappoport, Z.; Stirling, C. J. M.; Eds.; John Wiley & Sons, 1988. (b) Posner, G. H. Acc. Chem. Res. 1987, 20, 72-87. (c) Solladié, G. Synthesis 1981, 185-196. (d) Walker, A. J. Tetrahedron: Asymmetry, 1992, 3, 961-968.
- For examples of diastereoselective Diels-Alder reactions of unsaturated sulfoxides, see: (a) Ronan, B.; Kagan, H. B. Tetrahedron: Asymmetry 1992, 3, 115-122. (b) Arai, Y.; Matusi, M.; Koizumi, T.; Shiro, M. J. Org. Chem. 1991, 56, 1983-1985. (c) Carreño, M. C.; García-Ruano, J. L.; Urbano, A. Tetrahedron Lett. 1989, 30, 4003-4006. (d) Fisher, M. J.; Overman, L. E. J. Org. Chem. 1988, 53, 2630-2634. (e) Posner, G. H.; Harrison, W. J. Chem. Soc., Chem. Commun. 1985, 1786-1787. (f)

- Overman, L. E.; Petty, C. B.; Ban, T.; Huang, G. T. J.Am. Chem. Soc. 1983, 103, 6335-6337. (g) Evans, D. A.; Bryan, C. A.; Sins, C. L. J. Am. Chem. Soc. 1972, 94, 2891-2892.
- For enantiospecific lactonizations of enantiopure vinylsulfoxides, see: (a) Marino, J. P.; Laborde, E.;
 Paley, R. S. J. Am. Chem. Soc. 1988, 110, 966-968. (b) Marino, J. P.; Perez, A. D. J. Am. Chem.
 Soc. 1984, 106, 7643-7644. (c) Marino, J. P.; Fernández de la Pradilla, R. Tetrahedron Lett. 1985, 25,
 5381-5384. (d) Marino, J. P.; Bogdan, S.; Kimura, K. J. Am. Chem. Soc. 1992, 114, 5566-5572. (e)
 For the lactonization of a dienyl sulfoxide, see: Burke, S. D.; Shankaran, K.; Helber, M. J. Tetrahedron
 Lett. 1991, 32, 4655-4658.
- For recent synthetic methods employing enantiopure sulfoxides, see: (a) Pan, L. R.; Tokoroyama, T.
 Tetrahedron Lett. 1992, 33, 1469-1472. (b) Guillot, C.; Maignan C. *Tetrahedron Lett.* 1991, 32, 4907 4908. (c) Alexandre, C.; Belkadi, O.; Maignan, C. *Synthesis* 1992, 547-548. (d) Swindell, C. S.;
 Blase, F. R.; Eggleston, D. S.; Krause, J. *Tetrahedron Lett.* 1990, 31, 5409-5412. (e) García-Ruano, J.
 L.; Martín Castro, A. M.; Rodríguez, J. H. *Tetrahedron Lett.* 1991, 32, 3195-3198. (f) Marino, J. P.;
 Viso, A.; Fernández de la Pradilla, R.; Fernández, P. J. Org. Chem. 1991, 56, 1349-1351.
- (a) Solladié, G.; Maugein, N.; Morreno, I.; Almario, A.; Carreño, M. C.; García-Ruano, J. L.
 Tetrahedron Lett. 1992, 33, 4561-4562. (b) Girodier, L.; Maignan, C.; Rouessac, F. *Tetrahedron:* Asymmetry 1992, 3, 857-858. (c) Bonfand, E.; Gosselin, P.; Maignan, C. *Tetrahedron Lett.* 1992, 33, 2347-2348. (d) Aversa, M. C.; Bonaccorsi, P.; Giannetto, P.; Jafari, S. M. A.; Jones, D. N.
 Tetrahedron: Asymmetry 1992, 3, 701-704. (e) Solladié, G.; Ruiz, P.; Colobert, F.; Carreño, M. C.;
 García-Ruano, J. L. *Synthesis* 1991, 1011-1012. (f) Farina, V.; Hauck, S. I. *J. Org. Chem.* 1991, 56, 4317-4319.
- 7. (a) Mitchell, T. N.; Synthesis 1992, 803-815. (b) Stille, J. K.; Groh, B. L. J. Am. Chem. Soc. 1987, 109, 813-817. (c) Stille, J. K. Angew. Chem. Int. Ed. Engl. 1986, 25, 508-524 and references therein.
- (a) For trans-bromovinylsulfoxide 1a, see: Paley, R. S.; Ph.D. Thesis, University of Michigan 1986.
 Also, see ref. 4a. b) For (Z)-iodovinylsulfoxides 2 and 3, see: Fernández de la Pradilla, R.; Morente, M.; Paley, R. S. Tetrahedron Lett. 1992, 6101-6102.
- 9. These isomers are not available by any of the recent reports on the subject. (See ref. 6). Additionally, the presence of a strongly polarized sulfur-oxygen bond syn to the reactive center could interfere with the catalytic cycle and/or the stereochemical outcome of the process. Furthermore, the stereochemical stability of the products was also a matter of concern.
- Vinyl stannanes not commercially available were prepared by the method of: (a) Labadie, J. W.; Tueting, D.; Stille, J. K. J. Org. Chem. 1983, 48, 4634-4642. (b) 1. Beaudet, I.; Parrain, J.-L.; Quintard, J.-P. Tetrahedron Lett. 1991, 32, 6333-6336.
 Marek, I.; Alexakis, A.; Normant, J.-F. Tetrahedron Lett. 1991, 32, 6337-6340. (c) Soderquist, J. A.; Hsu, G. J.-H. Organometallics 1982, 1, 830-834, via the corresponding vinyl bromide. See: Seebach, D.; Neumann, H. Chem. Ber. 1974, 107, 847-853.
- 11. All new compounds have been satisfactorily characterized.